



The Hebrew University Center  
for Nanoscience & Nanotechnology



## Nano Seminar

# Two-dimensional oxides as new catalytic materials

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### **Abstract:**

Charging of supported clusters on oxide films can occur spontaneously for films of thickness of 1-2 nm via direct tunneling from the metal support [1,2]. This effect, which has been demonstrated for the case of gold atoms and clusters, can significantly modify the shape and chemical reactivity of supported metal clusters [1,2]. The presence of extra charge on the supported metal nanoparticle can have significant effects on catalytic reactions. Thin oxide films can also become active catalysts thanks to their structural flexibility [3] or nanoporosity [4]. Activation of supported metal clusters can be obtained also by doping an oxide film with transition metal atoms incorporated in the inner layers [5-8]. We will discuss the ability of Mo (Cr) impurities in a CaO (MgO) matrix to act as charge donors to adsorbed gold. Whereas  $\text{CaO}_{\text{Mo}}$  features a robust electron donor characteristic,  $\text{MgO}_{\text{Cr}}$  is electrically inactive due to the different stability of various oxidation states. Based on our findings, we develop general rules on how to optimize the electron-donor characteristics of doped oxide materials and thin films and how this can be used to tune the catalytic properties of supported nanoparticles.

[1] L. Giordano, G. Pacchioni, "Oxide films at the nanoscale as new catalytic materials", *Acc. Chem. Res.*, **44**, 1244 (2011).

[2] G. Pacchioni, H. J. Freund, "Electron transfer at oxide surfaces. The MgO paradigm: from defects to ultrathin films", *Chemical Reviews*, **113**, 4035 (2013).

[3] Q. Pan, X. Weng, M. Chen, L. Giordano, G. Pacchioni, C. Noguera, J. Goniakowski, S. Shaikhutdinov, H.-J. Freund, "Enhanced CO oxidation on metal/oxide interface: from ultrahigh vacuum to near-atmospheric pressures", *ChemCatChem*, **7**, 2620-2627 (2015).

[4] P. Schlexer, G. Pacchioni, R. Wlodarczyk, J. Sauer, "CO adsorption on a silica bilayer supported on Ru(0001)", *Surface Science*, **648**, 2-9 (2016).

[5] X. Shao, S. Prada, L. Giordano, G. Pacchioni, N. Nilius, H.-J. Freund, "Shape control of metal nanoparticles via doping of the oxide support: An STM and DFT study", *Angew. Chem. Int. Ed.*, **50**, 11525 (2011).

[6] F. Stavale, X. Shao, N. Nilius, H.-J. Freund, S. Prada, L. Giordano, G. Pacchioni, "Transition metal dopants and the electron donor characteristic of oxide surfaces", *J. Am. Chem. Soc.*, **134**, 11380 (2012).

[7] S. Prada, L. Giordano, G. Pacchioni, "Charging of gold atoms on doped MgO and CaO: identifying the key parameters by DFT calculations", *J. Phys. Chem. C*, **117**, 9943 (2013).

[8] S. Prada, L. Giordano, G. Pacchioni, "Nb-doped CaO: an efficient electron donor system", *J. Physics: Condens. Matter*, **26**, 315004(6) (2014).

### **Gathering & Refreshments at 10:50**

Please contact Alexandra Bannykh at 6584919 if you are interested in meeting the lecturer.

**Tuesday, Apr 4<sup>th</sup> 2017, 11:00 at the Seminar Hall**

Los Angeles Building, entrance floor.